



## Synthesis and Resolution of a Chiral Analogue of 2,2,6,6-Tetramethylpiperidine and of its Corresponding Nitroxide

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**Abstract :** The novel C<sub>2</sub> symmetric chiral amine *trans*-2,6-dimethyl-2,6-diphenylpiperidine has been synthesized via two successive nitron nucleophilic addition-oxidation sequences, followed by reduction of the intermediate nitroxide. Pure enantiomers have been obtained via resolution with mandelic acid.

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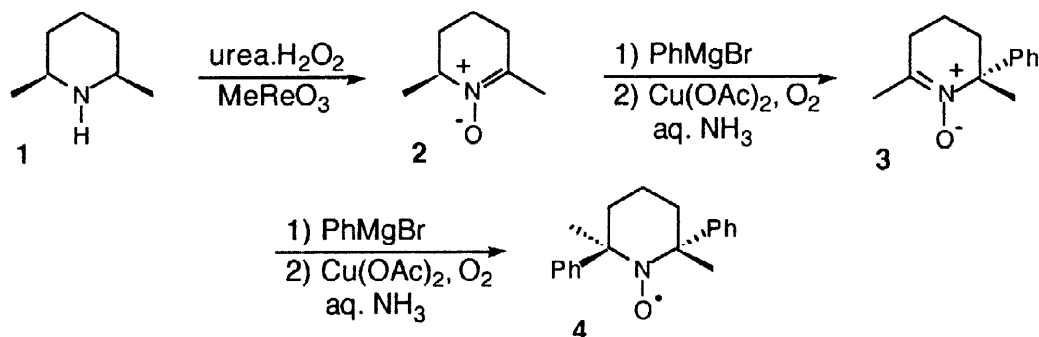
Highly sterically hindered C<sub>2</sub>-symmetric amines are of increasing importance for enantioselective synthesis and catalysis. For example, their lithium amides are used as chiral equivalents of LDA for enantioselective deprotonations.<sup>1</sup> Lithium 2,2,6,6-tetramethylpiperidide (LiTMP) or the corresponding magnesium amides (TMPMgX) are non nucleophilic strong bases which possess, in some cases, decisive advantages over LDA.<sup>2</sup> Chiral equivalents of LiTMP or TMPMgX are, therefore, valuable synthetic targets with interesting potential applications. Moreover, the stable nitroxides derived from the corresponding amines would themselves be of great interest: Recently chiral nitroxides have emerged as promising tools in various fields of chemistry, for example as enantioselective oxidation catalysts,<sup>3</sup> for the development of paramagnetic chiral liquid crystals,<sup>4</sup> in stereoselective coupling reactions with prochiral radicals<sup>5</sup> or in the control of living free-radical polymerisation processes.<sup>6,7</sup> We report here the synthesis of the novel, C<sub>2</sub> symmetric chiral amine *trans*-2,6-dimethyl-2,6-diphenylpiperidine **5** (Scheme 1) and of the corresponding nitroxide **4**, as well as the obtention of optically active **5** and **4** via resolution of amine **5**.

Our synthetic approach is based on tandem nucleophilic addition-oxidation sequences on nitron **2** obtained by oxidation of piperidine **1**, allowing the sequential introduction of the phenyl substituents. Such an approach has originally been developed by Keana for the synthesis of various pyrrolidinylnitroxides. He has demonstrated that it allows a high stereochemical control of the newly created stereocenters.<sup>8</sup> Müllen et al.<sup>9</sup> have applied this methodology for the synthesis of *trans*-2,5-dimethyl-2,5-diphenylpyrrolidine-1-oxyl and have carried out its resolution by preparative chiral HPLC. Recently, we have described an enantioselective approach to the same compound and its reduction to the corresponding amine.<sup>10</sup> To our knowledge, Keana's methodology has not been applied previously in the piperidine series.

As outlined in scheme 1, commercial *cis*-2,6-dimethylpiperidine **1** was first oxidized to nitron **2** by the complex urea-hydrogen peroxide catalysed by methyltrioxorhenium.<sup>11</sup> Nitron **2** was reacted with phenylmagnesium bromide in THF, followed by oxidation of the crude reaction product with molecular oxygen in the presence of ammoniacal cupric acetate,<sup>8</sup> leading to nitron **3**. As nitron **3** has a low stability it was used

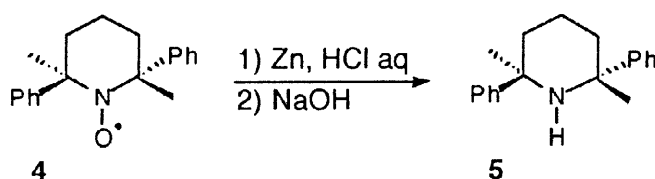
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without purification: the same Grignard addition-oxidation sequence furnished nitroxide **4**, isolated as yellow crystals after column chromatography purification (yield 8% from **2**, unoptimized).<sup>12</sup> Only one diastereomer was obtained. X-ray crystallography revealed the trans relationship of the methyl and of the phenyl groups respectively (Fig. 1).<sup>13</sup>



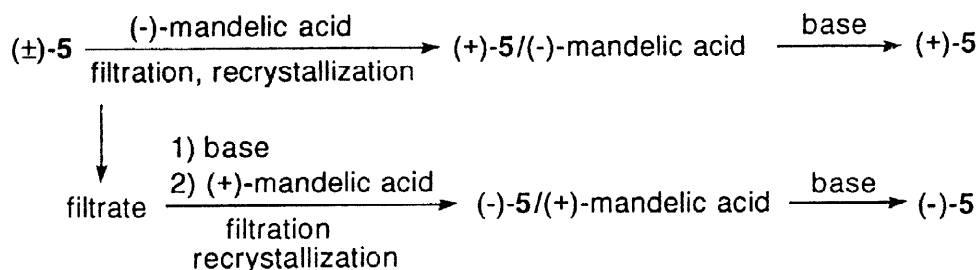
**Scheme 1** Synthesis of (+)-*trans*-2,6-dimethyl-2,6-diphenylpiperidin-1-oxyl **4**

Nitroxide **4** was next reduced by treatment with zinc powder in aqueous hydrochloric acid, followed by alkaline work-up.<sup>14</sup> Amine **5** was isolated with a 86% yield (Scheme 2).<sup>15</sup>



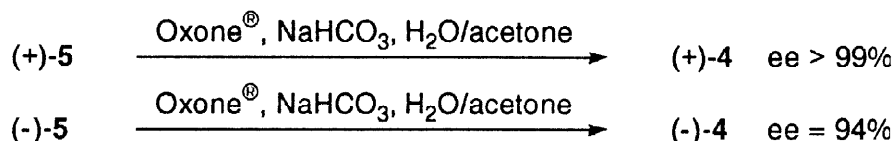
**Scheme 2** Reduction of nitroxide **4** to amine **5**

(±)-Amine **5** has been resolved by fractional crystallization of its diastereomeric salts formed with mandelic acid (Scheme 3): (*R*)-(-)-mandelic acid gives a crystalline salt with (±)-**5** from ether/ethanol, 3:1. After one recrystallization of this salt from ethyl acetate and alkaline treatment, (+)-**5** was obtained in 26% yield (theoretical yield 50%), with  $[\alpha]_D^{21} +77$  (c 1.03, EtOAc). The filtrate from the isolation of the salt (+)-**5**/(-)-mandelic acid furnished, after alkaline treatment, amine **5** enriched in the (-) enantiomer. This sample was treated with 1 equivalent of (*S*)-(+)-mandelic acid, furnishing the crystalline salt (-)-**5**/(+)-mandelic acid from ether/ethanol, 3:1. One recrystallization of this salt from ethyl acetate, followed by alkaline work-up gave amine (-)-**5** in 24% yield,  $[\alpha]_D^{21} -82$  (c 1.18, EtOAc).



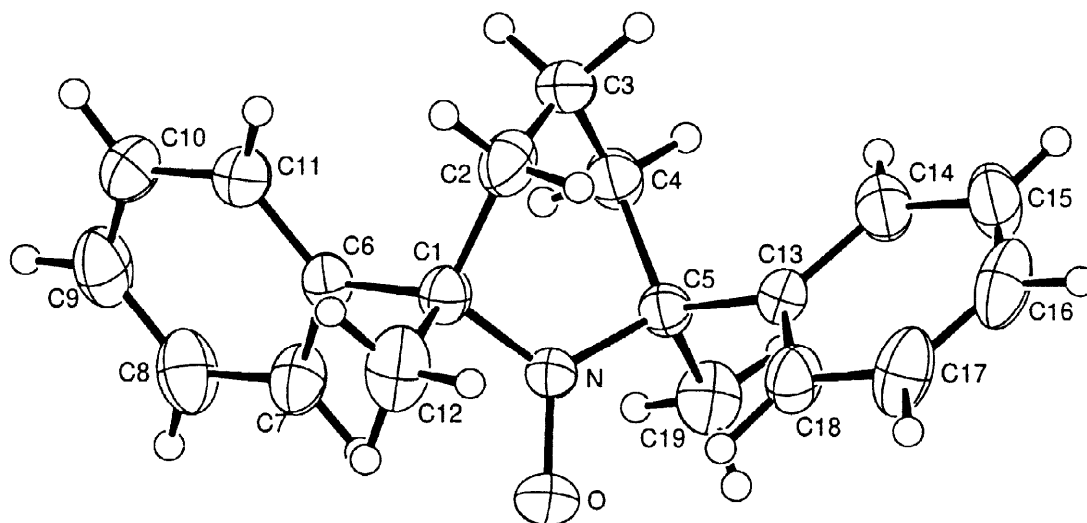
**Scheme 3** Resolution of racemic **5**

Both enantiomers of amine **5** were finally reoxidized into enantiomeric nitroxides **4**, using Oxone<sup>®</sup> (Scheme 4).<sup>16</sup> (-)-**5** (with  $[\alpha]_{\text{D}}^{21} -82$ ) gave nitroxide (-)-**4** in 85% yield :  $[\alpha]_{\text{D}}^{21} -140$  (c 1.11, EtOAc); ee > 99%.<sup>17</sup> Enantiomeric purity of (-)-**4** was measured after column chromatography purification and without recrystallization. Therefore its value can be assumed to reflect the enantiomeric purity of amine (-)-**5**.<sup>18</sup> Likewise amine (+)-**5** ( $[\alpha]_{\text{D}}^{21} +27$ ) was oxidized into (+)-**4** with  $[\alpha]_{\text{D}}^{21} +132$  (c 1.09, EtOAc), ee = 94%. Since the molecule does not contain any heavy atoms the direct determination of the absolute configuration by X-ray was not possible. Efforts to crystallize derivatives of **5** containing heavy atoms or additional asymmetric centers of known configuration for X-ray diffraction are in progress.



**Scheme 4** Obtention of optically active **4**

In summary, we have demonstrated that the methodology originally developed by Keana for the synthesis of pyrrolidinylnitroxides can also be applied in the piperidine series, with a similar degree of diastereoselectivity. Thus, it offers a new access to C<sub>2</sub> symmetric chiral piperidines and to the corresponding nitroxides. Optically active amines and, subsequently, nitroxides can be obtained via resolution. Synthetic applications of both class of compounds are under investigation.



**Fig. 1** Crystal structure of nitroxide **4**. Selected bond lengths (Å) and angles (°): N-O 1.271(4), N-C(1) 1.482(4), N-C(5) 1.481(2), C(1)-N-O 119.5(1), C(5)-N-O 118.6(1), C(1)-N-C(5) 121.9. Dihedral angle between the planes of the two phenyl rings: 71.5(1)°.

## REFERENCES AND NOTES

- For recent examples of C<sub>2</sub> symmetric amines, see: Woldersdorf, M.; Kranich, R.; Schmalz, H. G. *Tetrahedron* **1997**, *53*, 7219-7230. For applications of C<sub>2</sub> symmetric amines in synthesis see references cited therein.
- See for example: Misumi, A.; Iwagana, K.; Furuta, K.; Yamamoto, H. *J. Am. Chem. Soc.* **1985**, *107*, 3343-3345; Dougherty, C. M.; Olofson, R.A. *Org. Synth. Coll. Vol. 6* **1988**, 571-575 and references cited therein; Yanagisawa, A.; Yasur, K.; Yamamoto, H. *J. Chem. Soc., Chem. Commun.*, **1994**, 2103-2104; Schlecker, W.; Huth, A.; Ottow, E. *J. Org. Chem.*, **1995**, *60*, 8414-8416.
- Ma, Z.; Huang, Q.; Bobbitt, J.M. *J. Org. Chem.* **1993**, *58*, 4837-4843; Rychnovsky, S.D.; Mc Leron, T.L.; Rajapakse, H. *J. Org. Chem.* **1996**, *61*, 1194-1195.
- Tamura, R.; Susuki, S.; Azuma, N.; Matsumoto, A.; Toda, F.; Kamimura, A.; Hori, K. *Angew. Chem. Int. Ed. Engl.* **1994**, *33*, 878-879; Tamura, R.; Susuki, S.; Azuma, N.; Matsumoto, A.; Toda, F.; Ishii, Y. *J. Org. Chem.* **1995**, *60*, 6820-6825.
- Braslau, R.; Burrill II, L.C.; Mahal, L.K.; Wedeking, T. *Angew. Chem. Int. Ed. Engl.* **1997**, *36*, 237-238.
- Puts, R.D.; Sogah, D.Y. *Macromolecules* **1996**, *29*, 3323-3325.
- For a general review on the synthesis and applications of optically active nitroxides, see: Naik, N.; Braslau, R. *Tetrahedron* **1998**, *54*, 667-696.
- Lee, T.D.; Birrel, G.B.; Keana, J.F.W. *J. Am. Chem. Soc.* **1978**, *100*, 1618-1619; Keana, J.F.W.; Seyedrezai, S.E.; Gaughan, J. *J. Org. Chem.* **1983**, *48*, 2644-2647; Keana, J. F. W.; Cuomo, J.; Lex, L.; Seyedrezai, S.E. *J. Org. Chem.* **1983**, *48*, 2647-2654; Keana, J.F.W.; Prabhu, V.S. *J. Org. Chem.* **1986**, *51*, 4300-4301.
- Benfaremo, N.; Steenbock, M.; Klapper, M.; Müllen, K.; Enkelmann, V.; Cabrera, K. *Liebigs Ann.* **1996**, 1413-1415.
- Einhorn, J.; Einhorn, C.; Ratajczak, F.; Gautier-Luneau, I.; Pierre, J.L. *J. Org. Chem.* (in press).
- Goti, A.; Nanelli, L. *Tetrahedron Lett.* **1996**, *37*, 6025-6028.
- Nitroxide **4** is stable and could be stored at room temperature for several weeks without noticeable decomposition. It melts at 127 - 127.5°C after recrystallization from hexane. Its ESR spectrum exhibits a characteristic triplet hyperfine structure ( $g = 2.0066$ ,  $a_N = 14.06$ ). The <sup>1</sup>H-NMR spectrum consists of only unresolved, very broad signals. IR (KBr) 3088, 3063, 3026, 2979, 2959, 2945, 1494, 1448, 1371, 1266, 1029 cm<sup>-1</sup>. UV-vis 242 nm ( $\epsilon = 2705$ ), 446 nm ( $\epsilon = 2$ ). DCI-MS (NH<sub>3</sub> + isobutane) :  $m/z$  (%) = 281 (100), 266 (9), 250 (15). Anal. calcd for C<sub>19</sub>H<sub>22</sub>NO : C, 81.39; H, 7.91; N, 5.00. Found: C, 81.59; H, 7.83; N, 5.04.
- X-ray crystal data for racemic nitroxide **4**: C<sub>19</sub>H<sub>22</sub>NO,  $M_r = 280.39$ , monoclinic, space group P2<sub>1</sub>/c (No. 14),  $a = 8.843(2)$ ,  $b = 18.315(4)$ ,  $c = 10.731(3)$  Å,  $\beta = 114.17(2)^\circ$ ,  $V = 1585.6(7)$  Å<sup>3</sup>,  $Z = 4$ ,  $\rho_{\text{calcd}} = 1.174$  g cm<sup>-3</sup>,  $R = 0.034$ ,  $R_w = 0.038$ , for 1306 reflections with  $I > 3\sigma(I)$  and 191 variables. Data were recorded on a Nonius CAD4 diffractometer with Mo-K $\alpha$  radiation. The structure was solved by direct methods (Sir 92) and refined using full-matrix least squares methods. Hydrogen atoms were included, but not refined. Full data will be deposited at the Cambridge Crystallographic Data Center (CCDC). For a comparison with the X-ray data of the homologous pyrrolidinylnitroxide, see ref. 9.
- Rosantsev, E.G.; Sholle, V.D. *Synthesis* **1971**, 401-414.
- Selected data for amine **5** : Colorless oil; b.p. 165°C (0.15 mm). IR (neat) 3308, 3092, 3067, 3024, 2931, 2869, 1603, 1492, 1239, 1078, 1035 cm<sup>-1</sup>; <sup>1</sup>H-NMR (250 MHz)  $\delta$  (ppm) 1.03 (s, 6H) 1.54 (bs, 1H), 1.61 - 1.71 (m, 2H), 1.81 - 1.88 (m, 2H), 2.10 - 2.20 (m, 2H), 7.18 - 7.67 (m, 10H); <sup>13</sup>C NMR (62.5 MHz)  $\delta$  (ppm) 18.6, 33.1, 36.6, 54.9, 125.5, 125.9, 127.8, 150.4; DCI-MS (NH<sub>3</sub> + isobutane) :  $m/z$  (%) = 266 (100), 250 (19.5).
- Brik, M.E. *Tetrahedron Lett.* **1995**, *36*, 5519-5520.
- Enantiomeric composition of optically active **4** have been determined by HPLC on a chiracel OD-H column, elution : isopropanol/hexane (1: 9), 0.5 mL min<sup>-1</sup>.
- Attempts to determine the enantiomeric purity of amine **5** directly by chiral HPLC or by using NMR techniques have failed so far.